# Material Safety Datasheet for Sodium Azide (NaAz)

Identification of the substance / preparation and of the company / undertaking		
Product name	Sodium azide	
Preparation	Lyophilised or reconstituted with water / buffer at a concentration of no greater than 0.02%.	
Supplier identification	Everest Biotech Ltd, Cherwell Innovation Centre, 77 Heyford Park, Upper Heyford, OX25 5HD.	
Composition / information on ingredients		
Chemical characterisation	Inorganic salt	
Chemical name	Sodium azide	
CAS number	26628-22-8	
EEC-No	247-852-1	

### Hazards identification

Very toxic if swallowed. Contact with acids liberates very toxic gas.

Irrigate thoroughly with water for at least 10 minutes. Seek medical advice.
Wash skin thoroughly with water. Remove contaminated clothing and wash before re-use. In severe cases, obtain medical attention.
Remove from exposure, rest and keep warm. In severe cases, seek medical advice.
Wash out mouth thoroughly with water and give plenty of water to drink. Seek medical advice.

# Fire fighting measures Special risks Suitable extinguishing media

May explode if heated. May evolve toxic fumes in fire. Not applicable.

### Accidental release measures

Wear appropriate protective clothing. Inform others to keep a safe distance.

Spread soda ash liberally over spillage.

If local regulations permit, mop up cautiously with plenty of water and run to waste, diluting greatly with running water.

Otherwise transfer to container and arrange removal by disposal company. Wash site of spillage thoroughly with water.

Handling and storage Handling	Avoid prolonged contact with copper or lead, especially in drainage systems or mercury and other heavy metals which may result in the formation of explosive azides. Under no circumstances eat, drink or smoke while
	handling this material. Wash hands thoroughly after working with this material. Contaminated clothing should be removed and washed before re-use.

Store at 4°C Keep container closed and protected from direct sunlight and moisture. Store away from combustible materials.

## Exposure controls / personal protection

As appropriate to quantity handled. Respirator Ventilation Gloves Eve protection Other precautions

Dust respirator. Extraction hood. Rubber or plastic. Goggles or face shield. Plastic apron, sleeves, boots - if handling large quantities.

Physical and chemical properties	
Form	Solid
Colour	Pink/White
Odour	Odourless
Melting point	No data available.
Boiling temperature	No data available.
Density	No data available.
Vapour pressure	No data available.
Solubility in water	Very soluble.
Flash point	
Explosion limits	No data available.
Ignition temperature	No data available.

### Stability and reactivity

Stable unless heated.

Slow reaction at ambient temperature unless water contains dissolved carbon dioxide. Decomposes violently with bromine or chromyl chloride.

Contact with acids liberates highly toxic gas: forms readily detonatable salts with many metals, particularly heavy metals.

# Toxicological information

After ingestion, irritation of mucous membranes in the mouth, pharynx, oesophagus and gastrointestinal tract. Danger of skin absorption.

Systemic effect

**Toxic effects Further data** 

Cardiovascular disorders, NS disorders, diarrhoea, tiredness. **Kidneys** 

LD50 27 mg/kg oral, rat. No evidence of carcinogenic properties. Evidence of mutagenic effects.

## Ecological information

The following applies to azides in general azides are toxic for aquatic organisms.

Biological effects	Fish <i>L. macrochirus</i> toxic from 1.5ppm upwards in 24h.
Approximate acute toxicity for lower organisms	5ma/l: < n

Approximate acute toxicity for lower organisms 5mg/l;</< p> Approximate toxicity for cold blooded animals 1mg/l (values stated for sodium azide).

# Disposal considerations

Chemical residues are generally classified as special waste, and as such covered by regulations which vary according to location. Contact your local waste disposal authority for advice, or pass to a chemical disposal company. Rinse out empty containers thoroughly before disposal.

Transportation information	
UN-No.	1687
ADR/RID	6.1,42'(b)
IMO	6.1/1687
IMDG class	6.1
ΙΑΤΑ	1687
Packaging group	II
Correct technical name	Sodium Azide

Regulatory information	
Labelling according to EEC directives	
Symbol	T+ Very toxic.
R-phrases	R28-32 Very toxic if swallowed. Contact with acids liberates very toxic gas.
S-phrases	S28-45 After contact with skin, wash immediately with plenty of water. In case of accident or if you feel unwell, seek medical advice immediately (show label where possible)
EEC-No.	247-852-1
UK exposure limits:	OES, Short term, mg/m <sup>3</sup> : 0.3 - Sodium azide (as NaN <sub>3</sub> )